		Lesson Pla	n Template			
Grade: 6			Subject: Physical Science			
	Worksheets, Chro	omebooks	Technology Needed: PHeT Simulation			
	al Strategies:		<b>Guided Practices and Concret</b>	e Application:		
	instruction	Peer teaching/collaboration/	Large group activity	Hands-on		
	practice	cooperative learning	Independent activity	Technology integration		
	c Seminar	Visuals/Graphic organizers	Pairing/collaboration	Imitation/Repeat/Mimic		
	ng Centers	PBL	Simulations/Scenarios	-		
Lecture		Discussion/Debate	Other (list)			
Techno		Modeling	Explain:			
integra			_			
Other (	list)					
Standard(s	)-		Differentiation			
,	•	S-PS1-4 Develop a model that				
		anges in particle motion,	Below Proficiency:			
•		a pure substance when	Students will be able to identify	the three phases of matter and		
•	•	•	how they change in particle mot			
tnermal er	nergy is added o	r removed.	taken away through an interactive			
011	`		record their understanding on a	paper worksheet. Students below		
Objective(s	s)		proficiency will likely need refe	rence to their notes, additional		
C. 1	11 1	1 / 12 (4 4 1	help from the teacher, and addit	ional help from the teacher aid.		
		understanding of the three phases				
		e changes in particle motion when	Above Proficiency:			
		y through an interactive	Students will demonstrate the th			
	Students will reco	ord their understanding on a paper	they change in particle motion v			
worksheet.			away through an interactive sim			
Bloom's Taxonomy Cognitive Level:			their understanding on a paper worksheet. Students above			
Bloom's Taxonomy Cognitive Level:			proficiency will be able to challenge themselves and work with			
Annly		the pressure simulation to understand how pressure effects				
Apply			volume and energy.			
			A 1: /E : D	e• •		
			Approaching/Emerging P			
				edge of the three phases of matter		
				motion when energy is added or		
			taken away through an interactive			
			record their understanding on a paper worksheet. Students approaching proficiency will likely need the reference to their			
		approaching proficiency will lik	ely need the reference to their			
			notes and minimal assistance from	om the teacher.		
- CI		• ()				
	Management- (g transitions, etc.)		Behavior Expectations- (system specific to the lesson, rules and			
~ .				·		
		ependently at their desks to		1 1 (0177) 1 11		
mınımalıze	any behavior issu	les.	Students are expected to be on t	÷		
			be listening and engaged when I am giving direct instruction.			
			They should be working and asking questions when they need			
			help during work time.			
Minutes		Procedu	res			
	Set-up/Prep:					
		g activity/ anticipatory Set – acces	s prior learning / stimulate inter	est /generate questions, etc.)		
	8 8 (-F	g , <sub>1</sub> ,		g 1 , <del></del> )		
	Review, what are	the three states of matter? (A. Solid	, liquid, gas) Which state of matte	r has the most energy? (A. gas)		
1 -	W/laiala akaka af	. 44 l 41				

Which state of matter has the least energy? (A. solid)

# 7 Explain: (concepts, procedures, vocabulary, etc.)

I will explain today's activity: Phase of Matter Simulation. I will pull up Google Classroom and show the students where they will find the link for today. Some important notes: They will click on STATES. I will go over how to use the simulation. They will want to turn temperature to Celsius. They can manually add/remove heat energy or they can click on the states of matter. When they are done they can also explore the pressure section as a challenge. I will pass out the paper portion of this practice and students will begin to explore.

Explore: (independent, concreate practice/application with relevant learning task -connections from content to real-life experiences, reflective questions- probing or clarifying questions)

Using the link below, students will explore the phases of matter simulation. (First they must make some predictions!) If students finish early they can work on:

- 1. The Chemical/Physical changes practices (Due Friday)
- 2. The Phase Change Practice (Going over Thursday)
- 3. Explore the pressure section of the simulation
- 4. Any late work

https://phet.colorado.edu/en/simulation/states-of-matter-basics https://phet.colorado.edu/en/simulation/states-of-matter-basics

2 Review (wrap up and transition to next activity):

The last two minutes students can pack up and put away materials to get ready for their next class.

# Formative Assessment: (linked to objectives) Progress monitoring throughout lesson- clarifying questions, check- in strategies, etc.

I will be monitoring student work using proximity to ask questions, redirect attention, and answer questions.

## **Summative Assessment (linked back to objectives)**

The completion of the simulation.

Additionally there will be a summative assessment at the end of the week that will cover phases of matter.

If applicable- overall unit, chapter, concept, etc.:

#### Reflection (What went well? What did the students learn? How do you know? What changes would you make?):

This lesson went well, the students were engaged during the direct instruction and were eager to answer the review questions I posed. The work time was also very productive, a few students completed the assignment during work time. I saw some students moving on to the pressure section of the simulation. If I were to change something from this lesson, I would have added an short activity or review to end the lesson after the students packed up their school supplies.

Name:		Dat	te:
			tes of Matter Simulation L
	Stat	tes of Matter Simula	ntion Lab
Before you o	pen the simulatio	on:	
PREDICT			
		ng what you think the moled	
matter, solid,	liquid, and gas. Wr	rite a sentence below each d	iagram predicting what th
the molecules	rivill be lilre		
	will be like.		
	will be like.		
	Solid	Liquid	Gas
Diagram of	T	Liquid	
	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of	T	Liquid	
Diagram of molecules	T	Liquid	
Diagram of molecules	T	Liquid	
Diagram of molecules  Sentence explaining	T	Liquid	
Diagram of molecules  Sentence explaining how	T	Liquid	
Diagram of molecules  Sentence explaining	T	Liquid	

ONCE YOU HAVE COMPLETED THIS PAGE, YOU MAY BEGIN THE SIMULATION.

Open the simulation.	You will fin	d it in a folde	r on your	desktop	labeled '	'States o	f Matter
Simulation."							

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			١пг	٠.

3. Use	e the menu	on th	e right sid	de of th	ie program	to select	Water	and	Solid.	Draw	and	describ	e
what	you see in	the sp	ace belov	N.									

Diagram	Description
4. Notice the thermometer at the top of the prograshowing?	am. What temperature scale is this thermometer
5. What happens to the water as you increase the	temperature?
	p
6. What is the melting/freezing point of water in	Celsius?
7. Add heat until the temperature is just below an How is water different below its melting point an	
8. Draw and describe what water looks like as a li	iquid.
Diagram	Description
	<u> </u>

9. What is the	boiling/condensation point	of water in Celsius?	
		t below and then just above t	
11. Draw and	describe what water looks l	ike as a gas.	
Diagram		Description	
happens when	n you add and remove heat i	ices listed in the menu on the from this substance. Use the Draw and describe its proper	buttons on the right to see
	Solid	Liquid	Gas
Diagram of molecules			
Sentence explaining how molecules are moving.			

ANALYZE:
13. How was this substance similar to water in each state of matter? How was it different?
14. Were your predictions (see p. 1) correct or incorrect? Explain.
15. Choose a substance other than water from the menu on the right side of the program. Use the slider to add and remove heat. Based on what the molecules do, figure out the approximate temperatures of the melting point and boiling point of this substance. (Hint: The temperatures given when you click solid, liquid, and gas are NOT the melting and boiling points.)  Substance:  Melting Point:  How did you figure it out?
Boiling Point:
How did you figure it out?